



# Non thermal plasma-assisted catalysis for methane abatement.

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# Introduction: CANMILK project

- **Carbon Neutral MILK (CANMILK):**
  - EU project.
  - 7 different organizations from across Europe.
- **Methane from dairy barns/meat farms.**
  - 5.4% of total greenhouse gases.
  - 28 times more potent than CO<sub>2</sub>.
  - Takes 10 to 12 years to decompose.
- **Goal: develop a simple, efficient and innovative technology based on plasma to capture methane emissions from dairy barns and convert them into carbon dioxide.**

For more information, please visit: [canmilk.eu](https://canmilk.eu)

## Consortium partners



**VTT Technical Research  
Centre of Finland**  
www.vttresearch.com  
Finland



**Valio**  
www.valio.com  
Finland



**University of Antwerp**  
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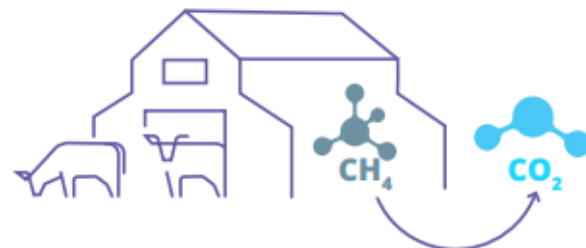
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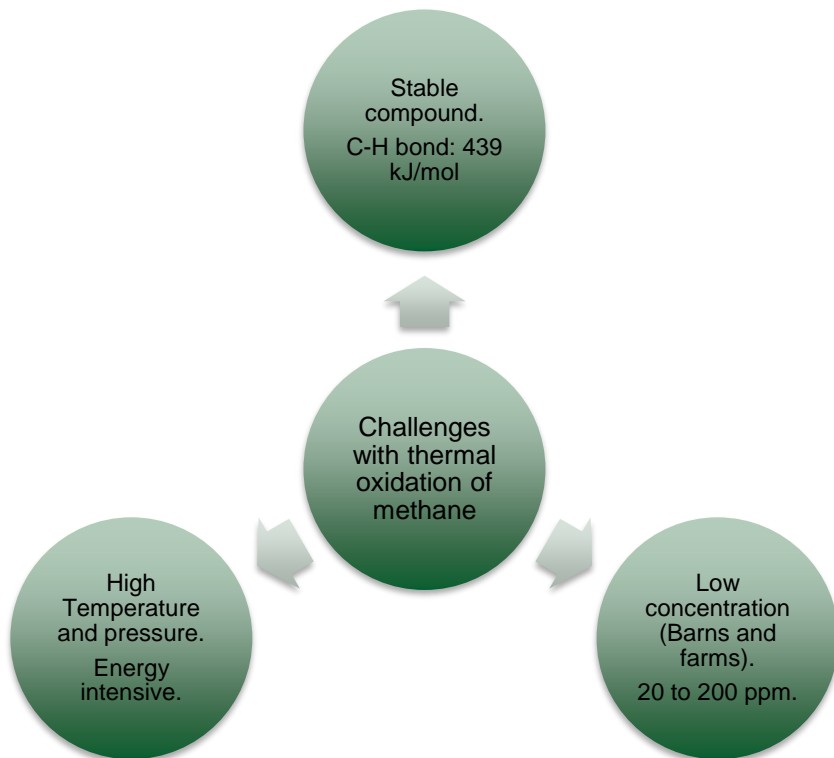


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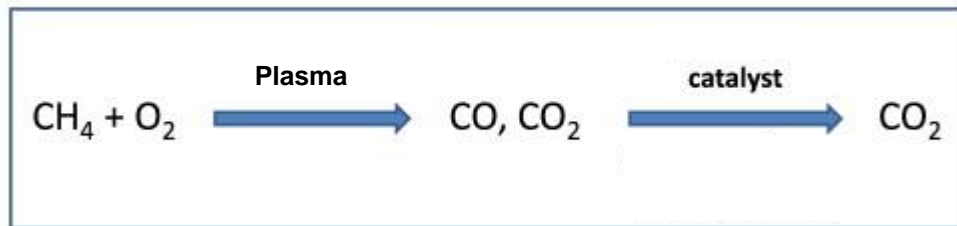


**Durham University**  
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United Kingdom





## Methane Complete Oxidation



Plasma oxidation scheme (Graphical abstract). Original from:  
Lee, H. et al. (2015) "Complementary effect of plasma–catalysis hybrid system on methane complete oxidation over non-PGM catalysts," Catalysis communications. DOI: <https://doi.org/10.1016/j.catcom.2015.07.001>

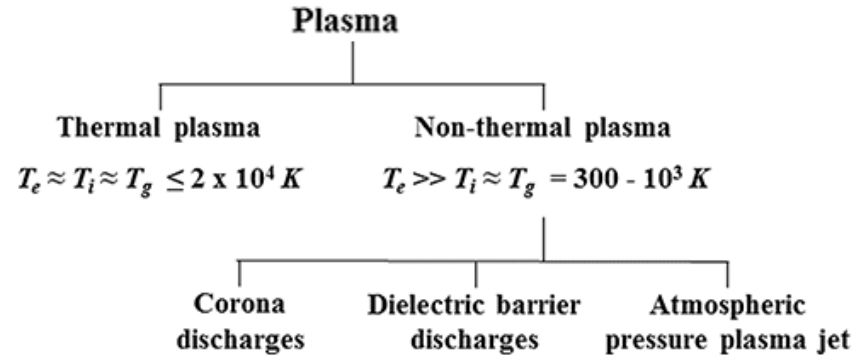
- ❑ Plasma produces highly reactive species, and radicals which can undergo gas phase reaction.
- ❑ However, gas phase plasma reaction is highly **nonselective**.
- ❑ The use of oxidation catalysts can improve **complete oxidation of methane into CO<sub>2</sub>**.



# Introduction: Plasma



- ❑ **Plasma:** fourth state of matter, partially or completely ionized.
- ❑ Types of plasma:
  - ❑ **Thermal plasma:** Thermal equilibrium, gas is heated to high temperature, using high energy source such as laser or microwave.
  - ❑ **Non-thermal plasma (NTP):** cold plasma or non-equilibrium plasma, electrons and ions are not in thermal equilibrium. generated at atmospheric pressure by passing the gas through the electric field.



Classification of plasma.

Original from: Lee, J. et al. (2017) "Use of Atmospheric Pressure Cold Plasma for Meat Industry," Korean Journal for Food Science of Animal Resources. Korean Society for Food Science of Animal Resources.

DOI: <https://doi.org/10.5851/kosfa.2017.37.4.477>



# Experimental set up

- ❑ Reactor: **Coaxial Dielectric barrier discharge (DBD) reactor.**
  - Two concentric quartz tubes.
  - Discharge length: 2 cm.
  - Discharge gap: 2 mm.
- ❑ Gases: 1 vol % CH<sub>4</sub> in air (0,99 % CH<sub>4</sub>, 20,89 % O<sub>2</sub> and 78,12 % N<sub>2</sub>) and volumetric flow of **200 mL/min.**
- ❑ Ambient conditions: atmospheric temperature and pressure.

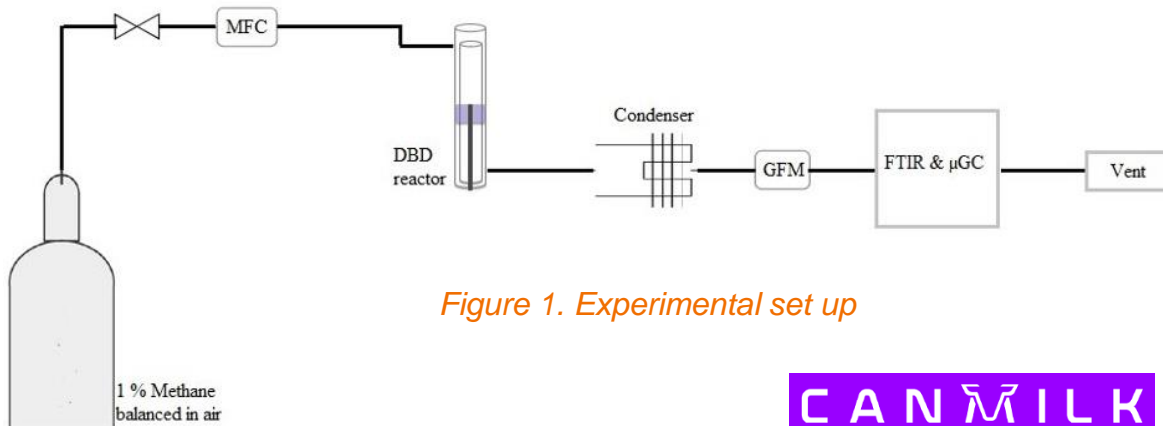
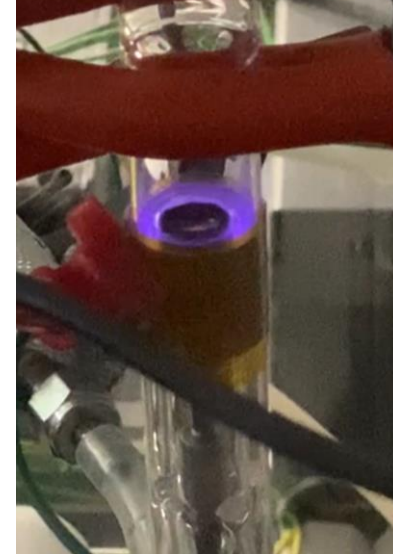


Figure 1. Experimental set up

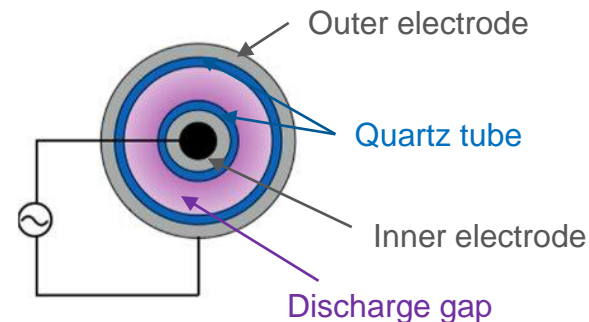


Figure 2. A co-axial DBD set up.

# Experimental set up

- Catalyst was placed inside the plasma discharge zone.
- 6 different catalysts were studied.

	Catalyst
1	$\gamma$ -Al <sub>2</sub> O <sub>3</sub>
2	1% Cu/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>
3	1% Pd/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>
4	1% Pt/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>
5	1% Pd – 0.5% Cu/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>
6	3% Pd/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>

- Catalysts were prepared by incipient wetness vacuum impregnation.



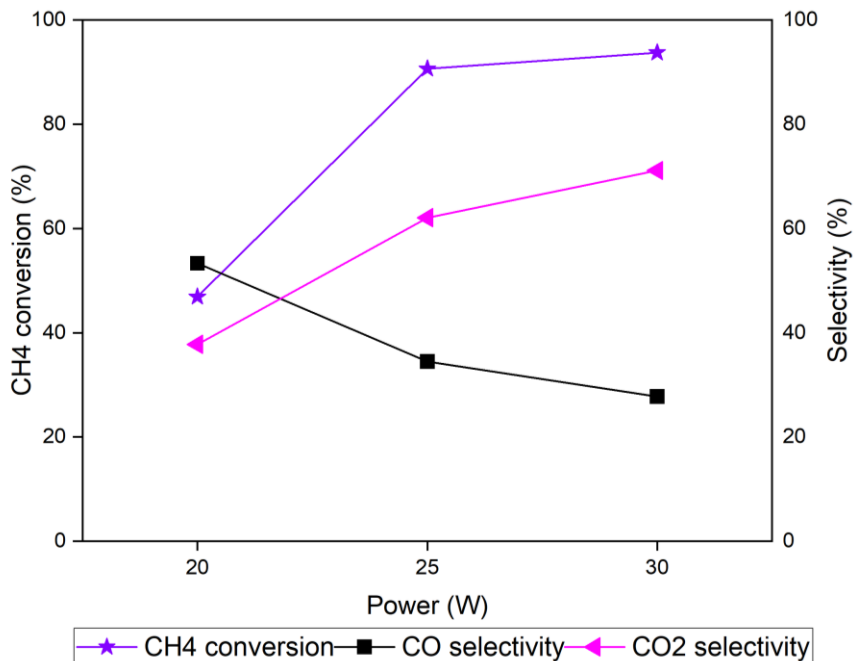
*In plasma catalyst reactor configuration. A. Side view, ground electrode is moved lower for better visualization B. Top view*

# Results: N<sub>2</sub> physisorption

<b>Fresh catalyst</b>	BET surface area (m <sup>2</sup> /g)	Pore volume (cm <sup>3</sup> /g)	Pore size (nm)
$\gamma$ -Al <sub>2</sub> O <sub>3</sub>	231	0,60	10
1 % Pd/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>	263	0,63	10
3 % Pd/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>	228	0,59	10
1 % Cu/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>	227	0,60	10
1 % Pd-0.5 % Cu/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>	225	0,60	10
1 % Pt/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub>	228	0,59	10
<b>Spent catalyst</b>			
1 % Pd/ $\gamma$ - Al <sub>2</sub> O <sub>3</sub>	259	0,63	10
1 % Cu/ $\gamma$ - Al <sub>2</sub> O <sub>3</sub>	224	0,60	10

- The BET surface area, pore volume, and pore size of all materials are in the same order of magnitude.
- 1% Pd/Al<sub>2</sub>O<sub>3</sub> catalyst had slightly higher BET surface area.
- The pore volume and BET surface are of the catalysts **did not change upon plasma exposure.**

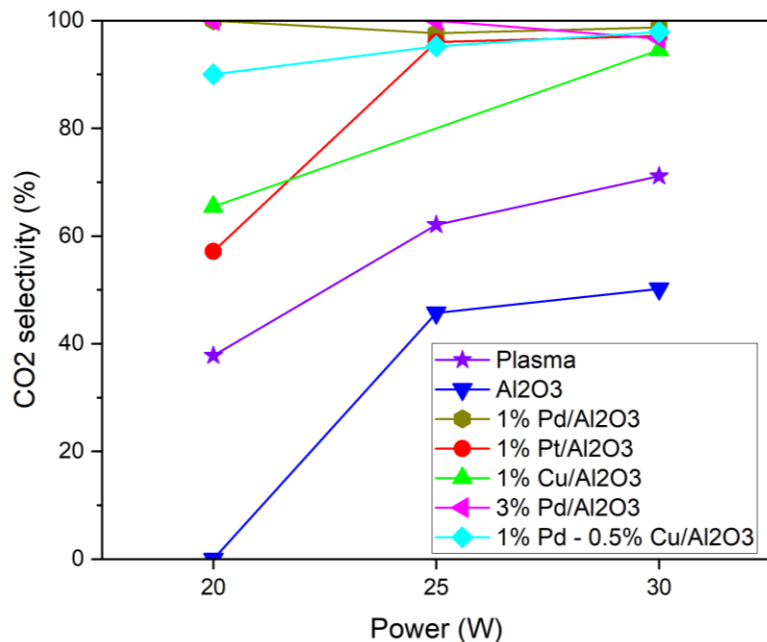
# Results: Plasma alone system



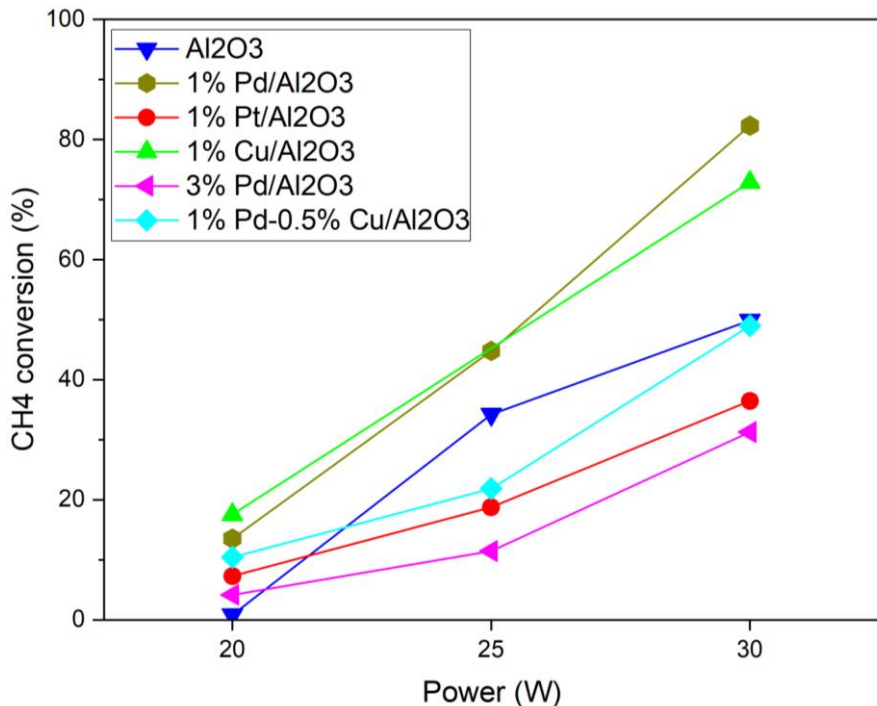
- ❑ Non-thermal plasma (NTP) achieved a 92% conversion of methane.
- ❑ The methane conversion increased with increase in plasma power.
- ❑ Methane was mostly converted into CO and CO<sub>2</sub>.
- ❑ The selectivity of CO<sub>2</sub> increased with higher plasma power while CO declined.



# Results: CO<sub>2</sub> selectivity



- ❑ The **complete oxidation of methane into carbon dioxide** was achieved in plasma-catalyst system.
- ❑ Pd based catalyst exhibited no CO selectivity across varying power levels.
- ❑ The Al<sub>2</sub>O<sub>3</sub> support demonstrated lower CO<sub>2</sub> selectivity compared to the plasma only system.



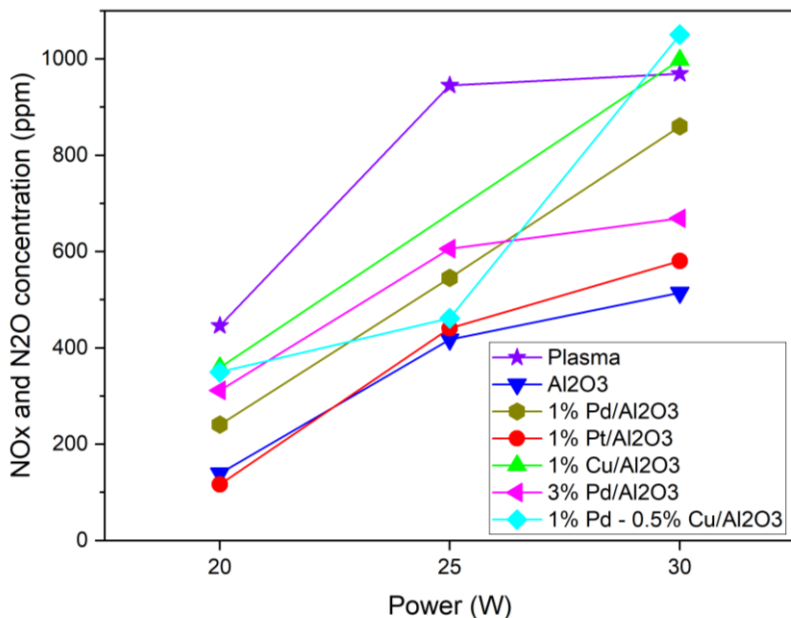
□ The order of activity is

**1 % Pd/Al<sub>2</sub>O<sub>3</sub> > 1 % Cu/Al<sub>2</sub>O<sub>3</sub> > γ-Al<sub>2</sub>O<sub>3</sub>**  
 > 1 % Pd – 0.5 % Cu/Al<sub>2</sub>O<sub>3</sub> > 1 % Pt/Al<sub>2</sub>O<sub>3</sub>  
 > 3 % Pd/Al<sub>2</sub>O<sub>3</sub>.

□ The conversion increased with increase in plasma power.

□ The Al<sub>2</sub>O<sub>3</sub> support did not show any activity at 20 W plasma power.

# Results: NO<sub>x</sub> and N<sub>2</sub>O formation



- ❑ The formation of undesirable NO<sub>x</sub> and N<sub>2</sub>O is unavoidable.
- ❑ The NO<sub>x</sub> formation is the function of catalytic activity.
- ❑ The NO<sub>x</sub> formation increases with increase in plasma power.
- ❑ The presence of catalyst did not suppress NO<sub>x</sub> and N<sub>2</sub>O formation.



## Conclusions

- ❑ NTP was able to achieve the highest methane conversion of 92% and CO<sub>2</sub> selectivity of 70%.
- ❑ 1% Pd/Al<sub>2</sub>O<sub>3</sub> catalyst showed the highest methane conversion of 84% and **CO<sub>2</sub> selectivity of 99%**.
- ❑ Combining NTP with catalysts proves to be an efficient method for **converting methane to carbon dioxide**, making it a promising technology for application in dairy barns and meat farms.

# Thank you for your attention!

For more information, please visit: [canmilk.eu](https://canmilk.eu)

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# References and credits:

- ❑ Slide 1: Picture by Ramon Salinero, from **unsplash.com**
- ❑ Slide 3: Graphical abstract: Lee, H. et al. (2015) "Complementary effect of plasma–catalysis hybrid system on methane complete oxidation over non-PGM catalysts," *Catalysis communications*.  
DOI: <https://doi.org/10.1016/j.catcom.2015.07.001>
- ❑ Slide 4: Thunderstorm picture taken by Melody P, from **unsplash.com**
- ❑ Slide 4: Northern lights picture taken by Jon Anders Dalan, from **unsplash.com**
- ❑ Slide 4: Classification of plasma. Original schematic from: Lee, J. et al. (2017) "Use of Atmospheric Pressure Cold Plasma for Meat Industry," *Korean Journal for Food Science of Animal Resources*. Korean Society for Food Science of Animal Resources. DOI: <https://doi.org/10.5851/kosfa.2017.37.4.477>